

# Iron supplement

<b>Iron supplement</b>	
Iron supplement from the late 19th and early 20th century	
<b>Clinical data</b>	
<b>Trade names</b>	Feosol, Feostat, Feratab, others
<b>Synonyms</b>	iron pills, iron salts, ferrous salts, ferric salts
<b>AHFS/Drugs.com</b>	Monograph
<b>Routes of administration</b>	by mouth, IV, IM
<b>ATC code</b>	B03A (WHO)
<b>Legal status</b>	
<b>Legal status</b>	US: OTC
<b>Identifiers</b>	
<b>CAS Number</b>	1332-96-3
<b>ChemSpider</b>	none

**Iron supplements**, also known as **iron salts** and **iron pills**, are a number of **iron** formulations used to treat and prevent **iron deficiency** including **iron deficiency anemia**.<sup>[1][2]</sup> For prevention they are only recommended in those with **poor absorption**, **heavy menstrual periods**, **pregnancy**, **hemodialysis**, or a diet low in iron.<sup>[2][3]</sup> Prevention may also be used in **low birth weight babies**.<sup>[2]</sup> They are taken by mouth, **injection into a vein**, or **injection into a muscle**.<sup>[2]</sup> While benefits may be seen in days up to two months may be required until iron levels return to normal.<sup>[4]</sup>

Common side effects include **constipation**, abdominal pain, dark stools, and **diarrhea**.<sup>[4]</sup> Other side effects, which may occur with excessive use, include **iron overload** and **iron toxicity**.<sup>[3][1]</sup> Ferrous salts used as supplements by mouth include **ferrous fumarate**, **ferrous gluconate**, **ferrous succinate**, and **ferrous sulfate**.<sup>[3]</sup> Injectable forms include **iron dextran** and **iron sucrose**.<sup>[3]</sup> They work by providing the iron needed for **making red blood cells**.<sup>[4]</sup>

Iron pills has been used medically since at least 1681 with an easy to use formulation being created in 1832.<sup>[6]</sup> It is on the [World Health Organization's List of Essential Medicines](#), the most effective and safe medicines needed in a [health system](#).<sup>[6]</sup> Ferrous salts are available as a [generic medication](#) and [over the counter](#).<sup>[1]</sup> The wholesale cost in the [developing world](#) is about 0.05 to 0.63 USD per month.<sup>[7]</sup> In the United States a typical month of treatment costs less than 25 USD.<sup>[1]</sup> [Slow release](#) formulations, while available, are not recommended.<sup>[2]</sup>

## Medical uses

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Iron supplements are used to treat [iron deficiency](#) and [iron-deficiency anemia](#).<sup>[8]</sup> parenteral irons can also be used to treat functional iron deficiency, where requirements for iron are greater than the body's ability to supply iron such as in inflammatory states. The main criterion is that other causes of anemia have also been investigated, such as vitamin B<sub>12</sub>/[folate](#) deficiency, drug induced or due to other poisons such as lead, as often the anaemia has more than one underlying cause.

Iron deficiency anemia is classically a microcytic, hypochromic anemia. Generally, in the UK oral preparations are trialled before using parenteral delivery.<sup>[9]</sup> unless there is the requirement for a rapid response, previous intolerance to oral iron or likely failure to respond. Intravenous iron may decrease the need for [blood transfusions](#) however increases the risk of infections when compared to oral iron.<sup>[10]</sup> A 2015 [Cochrane Collaboration](#) review found that daily oral supplementation of iron during pregnancy reduces the risk of maternal anaemia and that effects on infant and on other maternal outcomes are not clear.<sup>[11]</sup> Another review found that "intermittent regimens (of oral iron supplementation) produced similar maternal and infant outcomes as daily supplementation but were associated with fewer side effects". The quality of evidence was poor, though.<sup>[12]</sup>

## Athletes

Athletes may be at elevated risk of iron deficiency and so benefit from supplementation, but the circumstance vary between individuals and dosage should be based on tested [ferritin](#) levels, since in some cases supplementation may be harmful.<sup>[13]</sup>

## Side effects

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Side effects of therapy with oral iron are most often [diarrhea](#) or [constipation](#) and [epigastric](#) abdominal discomfort. Taken after a meal, side effects decrease, but there is an increased risk of interaction with other substances. Side effects are [dose-dependent](#), and the dose may be adjusted.

The patient may notice that his/her stools become black. This is completely harmless, but patients must be warned about this to avoid unnecessary concern. When iron supplements are given in a liquid form, teeth may reversibly discolor (this can be avoided through the use of a straw). Intramuscular injection can be painful, and brown discoloration may be noticed.

Treatments with [iron\(II\) sulfate](#) have higher incidence of adverse events than iron(III)-hydroxide polymaltose complex (IPC)<sup>[14][15][16]</sup> or iron bis-glycinate chelate.<sup>[17][18]</sup>

Iron overdose has been one of the leading causes of death caused by toxicological agents in children younger than 6 years.<sup>[19]</sup>

Iron poisoning may result in mortality or short-term and long-term morbidity.<sup>[20]</sup>

## Infection risk

Because one of the functions of elevated [ferritin](#) (an acute phase reaction protein) in acute infections is thought to be to sequester iron from bacteria, it is generally thought that iron supplementation (which circumvents this mechanism) should be avoided in patients who have active bacterial infections. Replacement of iron stores is seldom such an emergency situation that it cannot wait for any such acute infection to be treated.

Some studies have found that iron supplementation can lead to an increase in [infectious disease](#) morbidity in areas where bacterial infections are common. For example, children receiving iron-enriched foods have demonstrated an increased rate in [diarrhea](#) overall and enteropathogen shedding. Iron deficiency protects against infection by creating an unfavorable environment for bacterial growth. Nevertheless, while iron deficiency might lessen infections by certain pathogenic diseases, it also leads to a reduction in resistance to other strains of viral or bacterial infections, such as [Salmonella typhimurium](#) or [Entamoeba histolytica](#). Overall, it is sometimes difficult to decide whether iron supplementation will be beneficial or harmful to an individual in an environment that is prone to many infectious

diseases; however this is a different question than the question of supplementation in individuals who are already ill with a bacterial infection.<sup>[21]</sup>

Children living in areas prone for malarial infections are also at risk of developing anemia. It was thought that iron supplementation given to such children could increase the risk of malarial infection in them. A Cochrane systematic review published in 2016 found high quality evidence that iron supplementation does not increase the risk of clinical malaria in children.<sup>[22]</sup>

## Contraindications

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Contraindications often depend on the substance in question. Documented [hypersensitivity](#) to any ingredients and anemias without proper work-up (i.e., documentation of iron deficiency) is true of all preparations. Some can be used in iron deficiency, others require iron deficiency anaemia to be present. Some are also contraindicated in rheumatoid arthritis<sup>[23]</sup>

## Haemochromatosis

Individuals may be genetically predisposed to excessive iron absorption, as is the case with those with [HFE hereditary haemochromatosis](#). Within the general population, 1 out of 400 people has the homozygous form of this genetic trait, and 1 out of every 10 people has its heterozygous form.<sup>[24]</sup> Neither individuals with the homozygous or heterozygous form should take iron supplements.<sup>[24]</sup>

## Interactions

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Non-[heme](#) iron forms an insoluble complex with several other drugs, resulting in decreased absorption of both iron and the other drug. Examples include [tetracycline](#), [penicillamine](#), [methyldopa](#), [levodopa](#), [bisphosphonates](#) and [quinolones](#). The same can occur with elements in food, such as [calcium](#). Absorption of iron is better at a low pH (i.e. an acidic environment), and absorption is decreased if there is a simultaneous intake of antacids.

Many other substances decrease the rate of non-heme iron absorption. Examples are [tannins](#) from foods, such as [tea](#) and [saw palmetto](#), phytic acid and roughage.<sup>[25]</sup> Vegetarians and especially vegans are at increased risk of [iron deficiency](#) due to the combination of limited amounts of iron in the diet in a form that is poorly absorbed alongside compounds that further limit absorption.<sup>[citation needed]</sup>

Taken after a meal, there are fewer side effects but there is also less absorption because of interaction and pH alteration. Generally, an interval of 2–3 hours between the iron intake and that of other drugs seems advisable, but is less convenient for patients and can impact on compliance.

## History

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The first pills were commonly known as Blaud's pills,<sup>[26]</sup> which were named after [P. Blaud of Beaucaire](#), the French physician who introduced and started the use of these medications as a treatment for patients with [anemia](#).<sup>[27]</sup>

## Administration

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### By mouth

*"Proferrin" redirects here. For the pupil dilator, see [Profirin](#).*

Iron can be supplemented [by mouth](#) using various forms, such as [iron\(II\) sulfate](#). This is the most common and well studied soluble iron [salt](#) sold under brand names such as Feratab, Fer-Iron, and Slow-FE. It is in complex with [gluconate](#), [dextran](#), [carbonyl iron](#), and other salts. [Ascorbic acid](#), vitamin C, increases the absorption of non-heme sources of iron.<sup>[28]</sup>

Heme iron polypeptide (HIP) (e.g. Proferrin ES and Proferrin Forte) can be used when regular iron supplements such as ferrous sulfate or ferrous fumarate are not tolerated or absorbed. A clinical study demonstrated that HIP increased serum iron levels 23 times greater than ferrous fumarate on a milligram-per-milligram basis.<sup>[29]</sup>

Another alternative is *ferrous [glycine sulfate](#) or [ferroglycine sulfate](#)*, has less gastrointestinal side-effects than standard preparations such as iron fumarate.<sup>[30]</sup> It is unusual among oral preparations of iron supplements in that the iron in this preparation has very high oral bioavailability, especially in the liquid formulation. This option should be

evaluated before resorting to parenteral therapy. It is especially useful in iron deficiency anemia associated with [autoimmune gastritis](#) and [Helicobacter pylori](#) gastritis, where it generally has satisfactory effect.<sup>[31]</sup>

Since iron stores in the body are generally depleted, and there is a limit to what the body can process (about 2–6 mg/kg of body mass per day; i.e. for a 100 kg/220 lb man this is equal to a maximum dose of 200–600 mg/per day) without [iron poisoning](#), this is a chronic therapy which may take 3–6 months.<sup>[32]</sup>

Due to the frequent intolerance of oral iron and the slow improvement, parenteral iron is recommended in many indications.<sup>[33][34]</sup>

## By injection

Iron therapy (intravenously or intramuscular) is given when therapy by mouth has failed (not tolerated by the patient), oral absorption is seriously compromised (by illnesses, or when the patient cannot swallow), benefit from oral therapy cannot be expected, or fast improvement is required (for example, prior to elective surgery).<sup>[35]</sup> Parenteral therapy is more expensive than oral iron preparations and is not suitable during the first trimester of [pregnancy](#).<sup>[36]</sup>

There are cases where parenteral iron is preferable over oral iron. These are cases where oral iron is not tolerated, where the [haemoglobin](#) needs to be increased quickly (e.g. post partum, post operatively, post transfusion), where there is an underlying inflammatory condition (e.g. inflammatory bowel disease) or renal patients, the benefits of parenteral iron far outweigh the risks. In many cases, use of intravenous iron such as [ferric carboxymaltose](#) has lower risks of adverse events than a blood transfusion and as long as the person is stable is a better alternative.<sup>[37]</sup> Ultimately this always remains a clinical decision based on local guidelines, although National Guidelines are increasingly stipulating IV iron in certain groups of patients.<sup>[38][39]</sup>

Soluble iron salts have a significant risk of adverse effects and can cause toxicity due to damage to cellular macromolecules. Delivering iron parenterally has utilised various different molecules to limit this. This has included [dextrans](#), [sucrose](#), carboxymaltose and more recently Isomaltoside 1000.<sup>[40]</sup>

One formulation of parenteral iron is *iron dextran* which covers the old high molecular weight (trade name *DexFerrum*) and the much safer low molecular iron dextrans (tradenames including Cosmofer and Infed).<sup>[40]</sup>

Iron [sucrose](#) (trade names including *Venofer*) has an occurrence of allergic reactions of less than 1 in 1000.<sup>[41]</sup> A common side effect is taste changes, especially a [metallic taste](#), occurring in between 1 in 10 and 1 in 100 treated patients.<sup>[41]</sup> It has a maximum dose of 200 mg on each occasion according to the SPC, but it has been given in doses of 500 mg. Doses can be given up to 3 times a week.<sup>[42]</sup>

Iron [carboxymaltose](#) is marketed as *Ferinject*, *Injectafer*, and *Iropremand* in various countries.<sup>[43][44]</sup> The most common side effects are [headaches](#) which occur in 3.3%, and hypophosphatemia, which occurs in more than 35%.<sup>[45][46]</sup>

Iron [Isomaltoside 1000](#) (Trade name *Monofer*) is a newer formulation of parenteral iron that has a matrix structure that results in very low levels of free iron and labile iron. It can be given at high doses – 20 mg/kg in a single visit – no upper dose limit. This formulation has the benefit of giving a full iron correction in a single visit.<sup>[47][44]</sup>

## Follow-up

Follow-up is needed to ensure compliance and to detect adequate response to therapy. The interval of follow up can widely depend on both the method of administration, and the underlying pathology. For parenteral irons it is recommended that there be a period of 4 weeks before repeating blood test to allow the body to utilise the iron.<sup>[8]</sup> For oral iron, this can take considerably longer, so waiting 3 months may be appropriate.

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